



Road Salts and their Impact to our Source of Drinking Water: The Potomac River



***Salt Management in the Washington Region
Environmental and Transportation Perspectives***

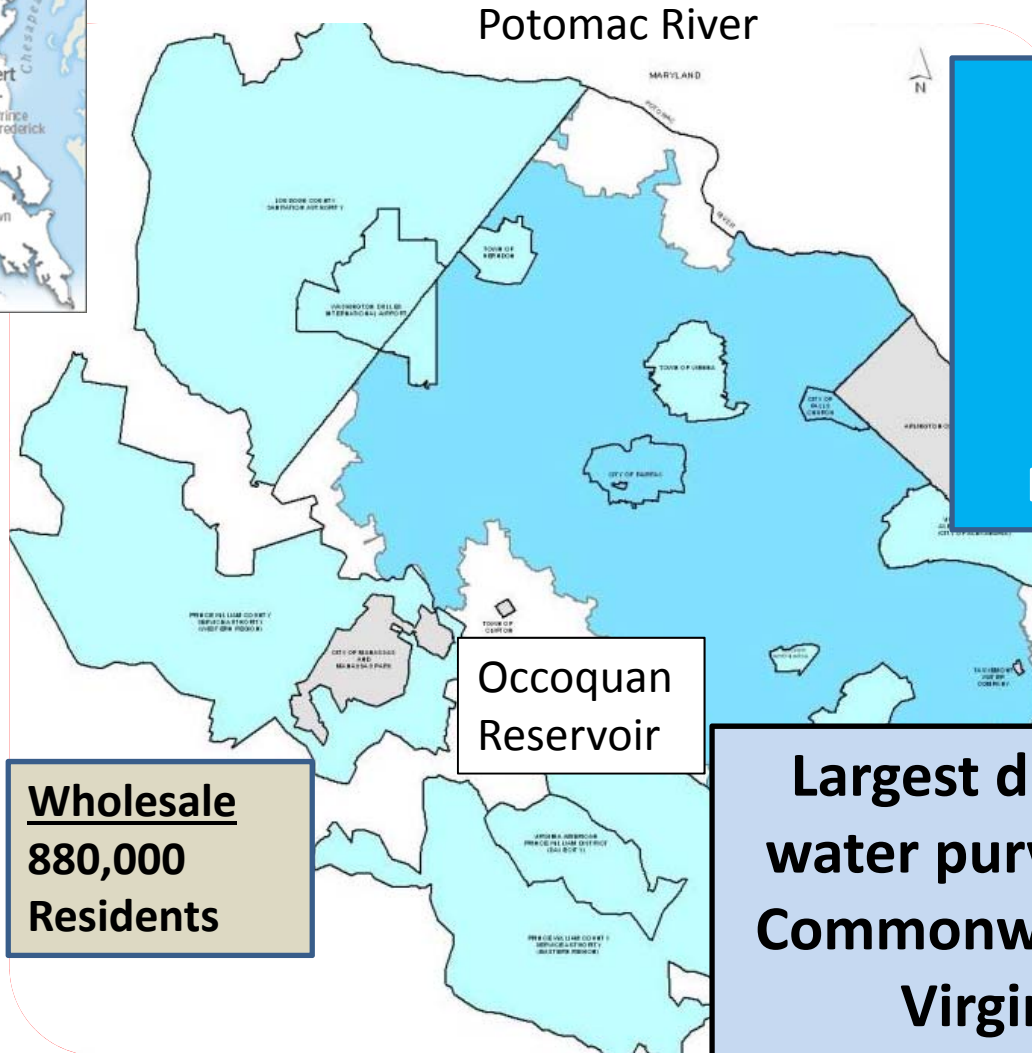
June 27, 2016

Fairfax Water

- Single-purpose not-for profit agency
- Serves almost 2 Million Virginians
- Nearly \$2 Billion in Infrastructure Assets
- Total Max Day Capacity = 375 MGD
- 10-Year CIP of \$720 M



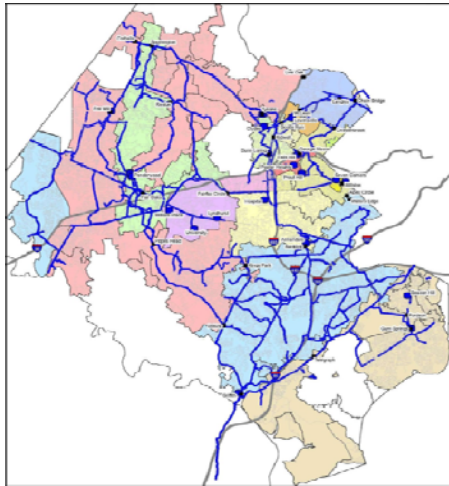
Fairfax Water



Retail
More than
1,075,000
Residents
and
11,000
Businesses

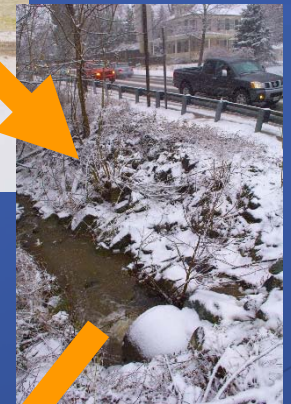
Wholesale
880,000
Residents

**Largest drinking
water purveyor in
Commonwealth of
Virginia**

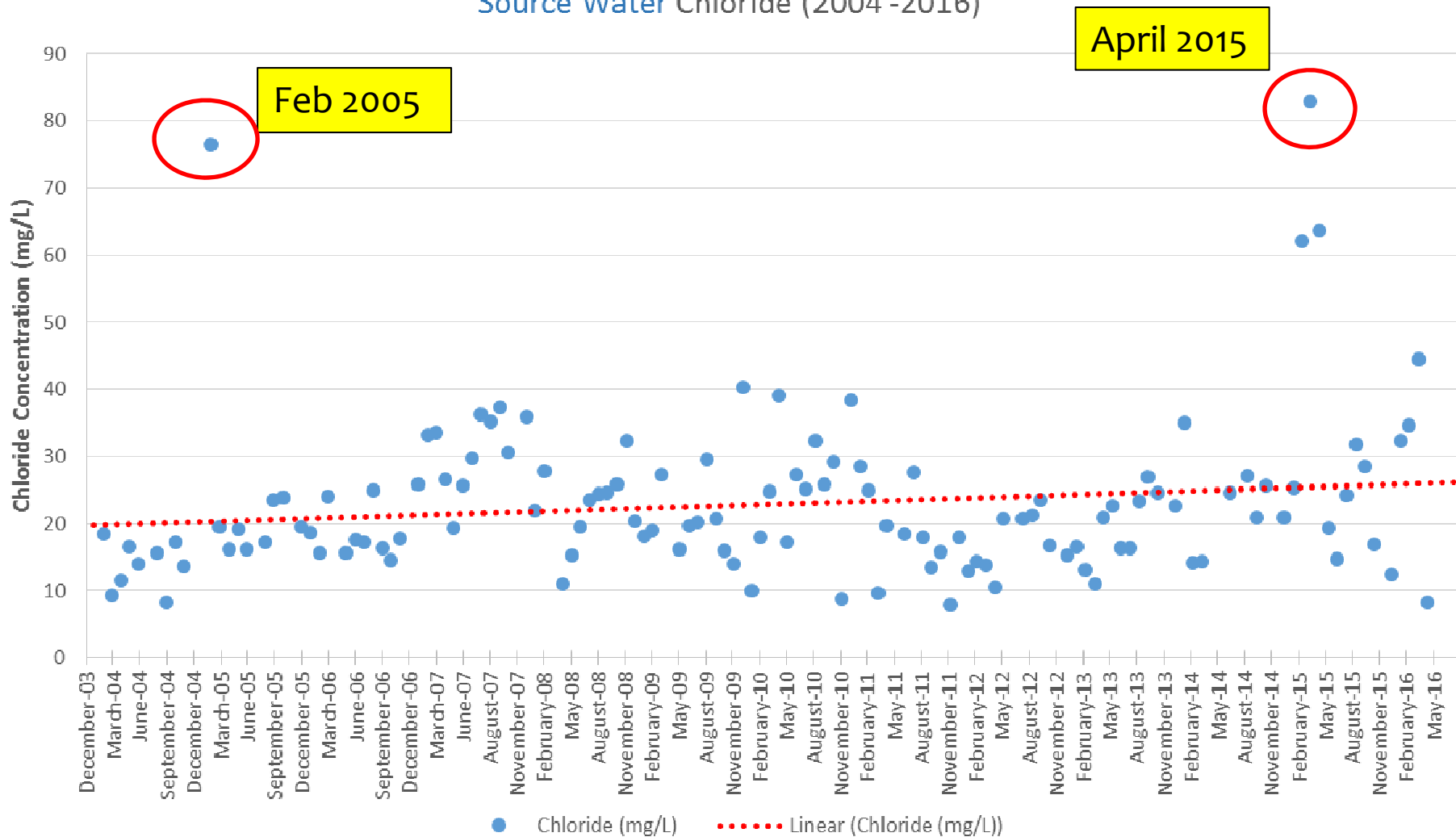


Water Quality Trends

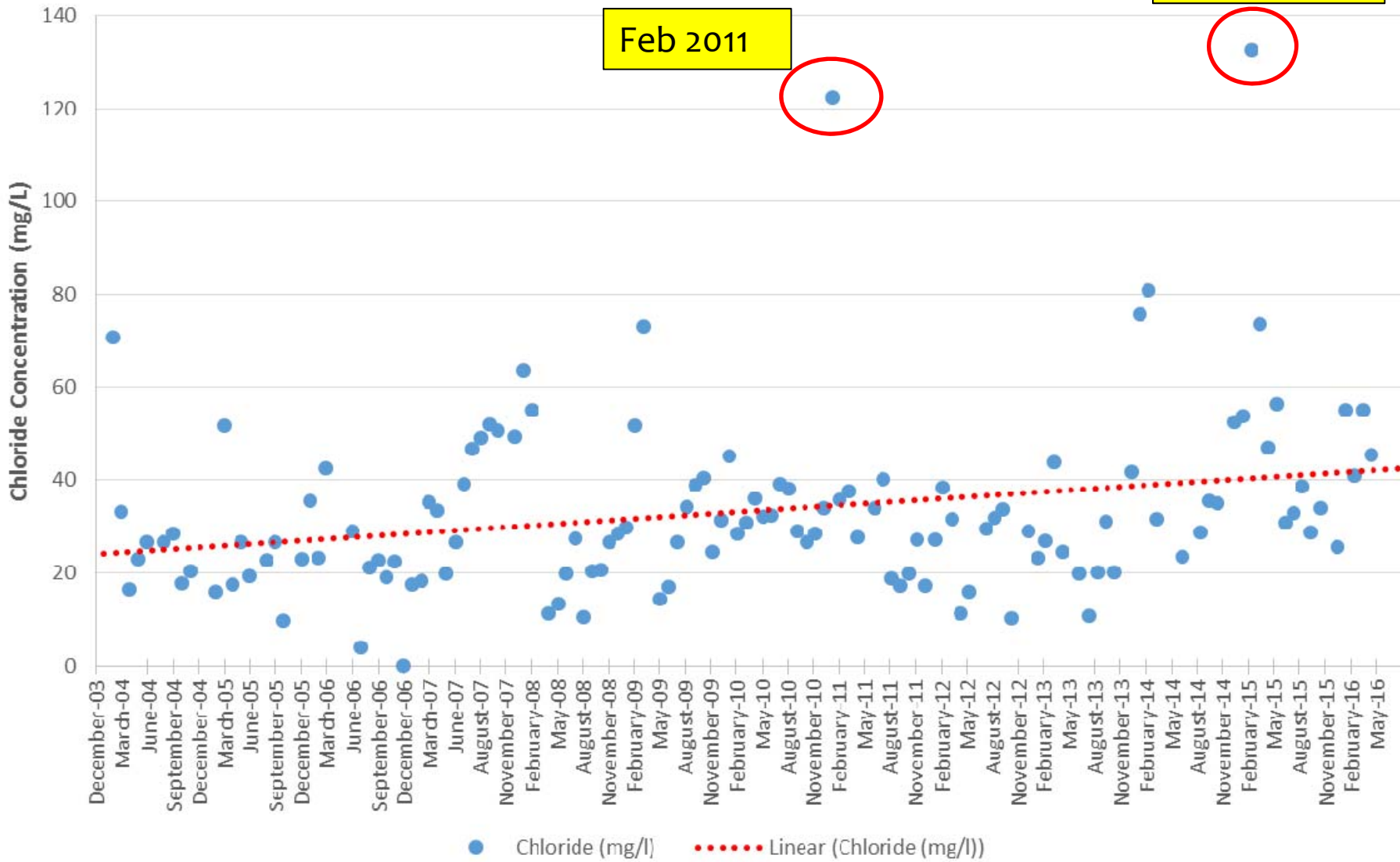
- Increasing Source Water Sodium and Chloride Concentrations
 - Seasonal Spikes
 - Secondary Drinking Water Standards and Advisory Levels
- Increasing trend for Bromide (Br⁻)
 - Commonly exists in the form of salts (sodium, potassium) from natural sources (rocks, soil)
- Potomac and Occoquan trends consistent with other regional and national studies



Fairfax Water Corbalis Treatment Plant Source Water Chloride (2004 -2016)



Fairfax Water Griffith Treatment Plant (Occoquan)
Source Water Chloride Concentration (2004 - 2016)



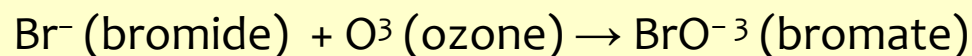
March 2015

Feb 2011

Bromide and Bromate



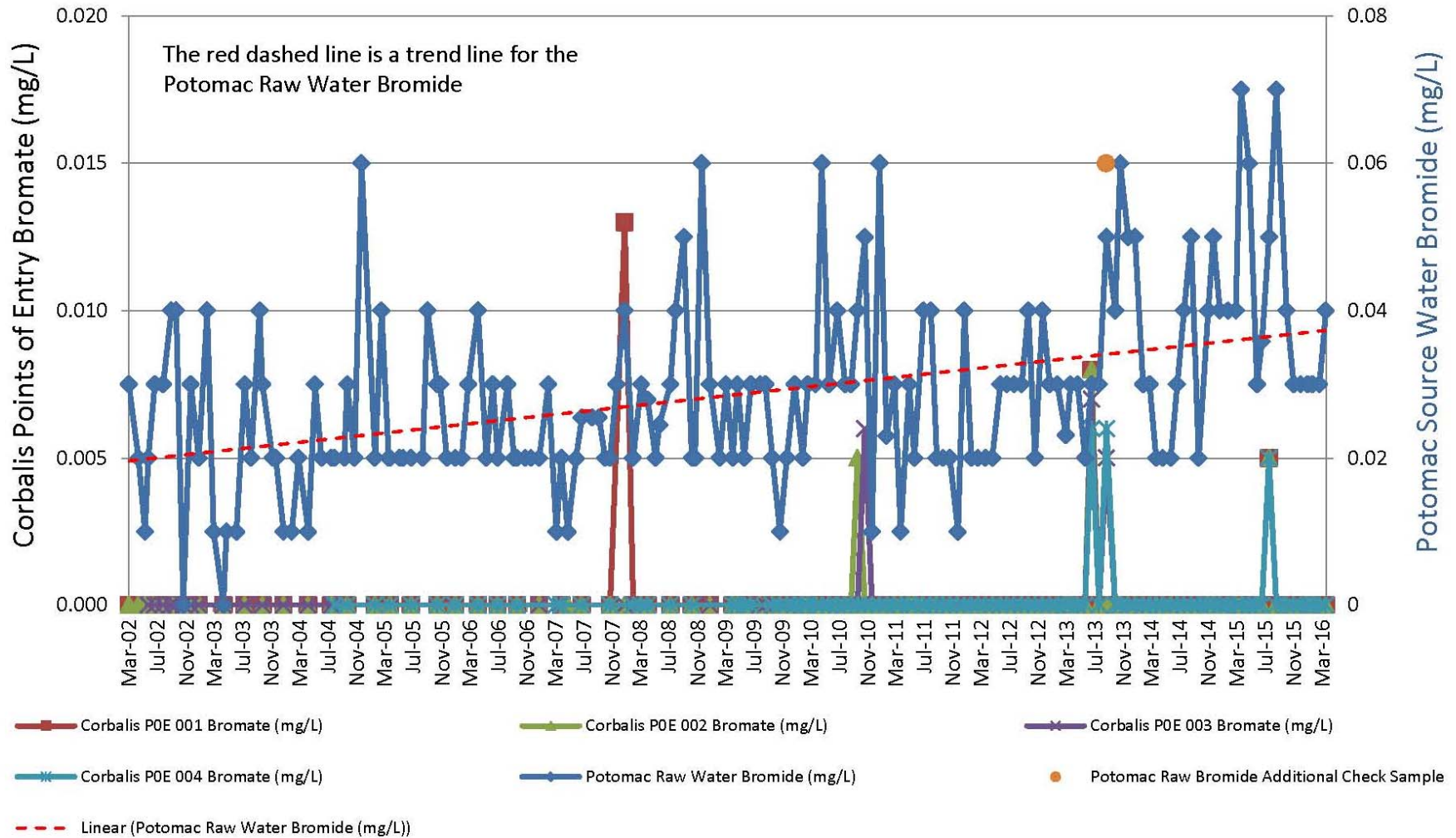
- Where present in the water, reactions between ozone or chlorine and naturally occurring organic matter form disinfection by products (DBPs) such as bromate



- Bromate Regulatory Compliance and Goals
 - Maximum Contaminant Limit (MCL)= 0.010 mg/L
 - Maximum Contaminant Limit Goal (MCLG) = zero

Fairfax Water Corbalis Treatment Plant

Source Water Bromide and Points of Entry Bromate



A plotted result of 0 indicates the result was below the laboratory's reporting level for the test. The reporting level for bromide is 0.01 mg/L. The reporting level for bromate is 0.005 mg/L (prior to July, 2007 the bromate reporting level was 0.01 mg/L).

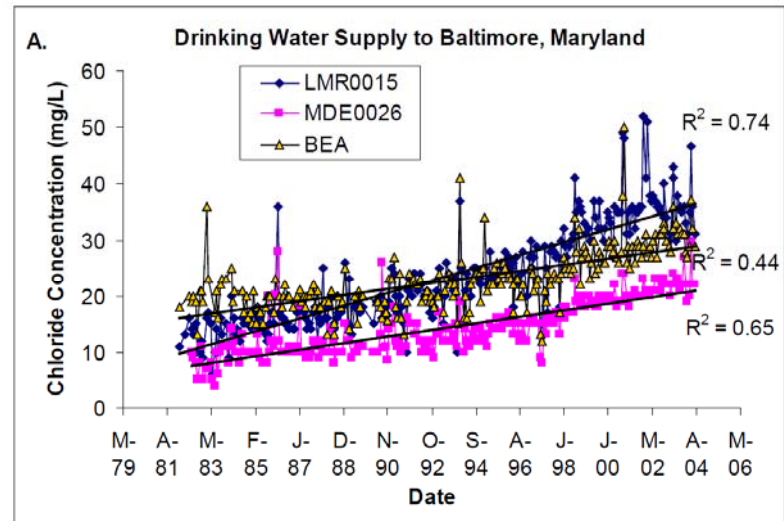
Consistent with other studies

**River Chloride Trends in Snow-Affected Urban Watersheds:
Increasing Concentrations Outpace Urban Growth Rate and Are
Common Among All Seasons**



Local, Regional, and National Impacts on
Aquatic Toxicity, Water-Quality and
Chloride Trends

Steven R. Corsi, Laura De Cicco, Michelle Lutz,
Robert Hirsch



Kaushal et al. (2005) *PNAS*

Other Water Quality Issues



Impact of Chloride: Sulfate Mass Ratio (CSMR) Changes on Lead Leaching in Potable Water

Subject Area: Infrastructure



Fairfax Water

1 **Title:** Effect of long-term changes in soil chemistry induced by road salt applications
2 on N-transformations in roadside soils.

3 Sophie M. Green*, Robert Machin and Malcolm S. Cresser
4 Environment Department, University of York, Heslington, York, YO10 5DD, UK

5
6 **Abstract:**

7 Of several impacts of road salting on roadside soils, the potential disruption of the
8 nitrogen cycle has been largely ignored. Therefore the fates of low-level ammonium-
9 N and nitrate-N inputs to roadside soils impacted by salting over an extended period
10 (decades) in the field have been studied. The use of road salts disrupts the
11 proportional contributions of nitrate-N and ammonium-N to the mineral inorganic
12 fraction of roadside soils. It is highly probable that the degree of salt exposure of
13 the soil, in the longer term, controls the rates of key microbial N transformation
14 processes, primarily by increasing soil pH. Additional influxes of ammonium-N to
15 salt impacted soils are rapidly nitrified therefore and, thereafter, increased leaching of
16 nitrate-N to the local waterways occurs, which has particular relevance to the Water
17 Framework Directive. The results reported are important when assessing the fate of
18 inputs of ammonia to soils from atmospheric pollution.

19
20 **"Capsule":** Road salting effects ammonification and nitrification in roadside soils.

21
22 **Keywords:** Road salt, ammonification, nitrification, roadside soils.

23
24 **1. Introduction:**

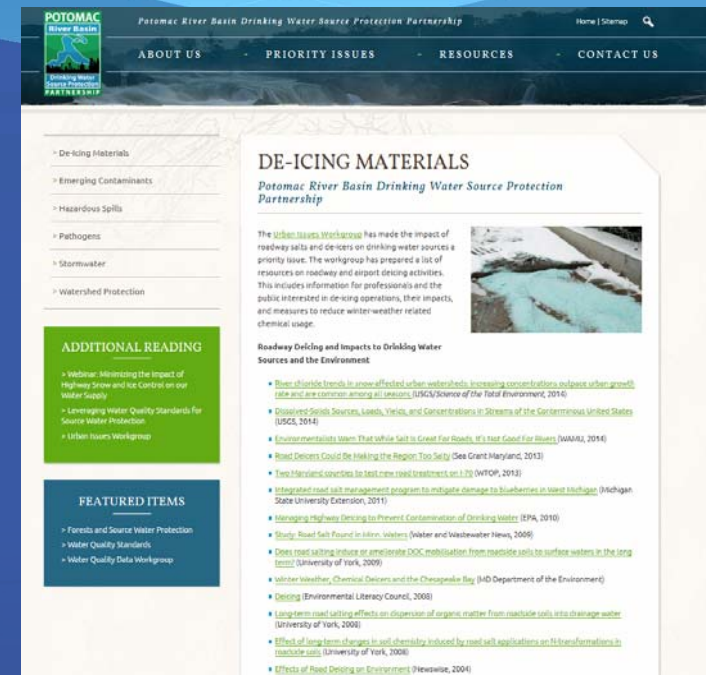
25 The application of deicing agents to roads has been widely practised in Europe and
26 North America during winter months since the 1960s to minimise the risk of accidents
27 due to ice and snow and to maintain traffic flow. Several different de-icing agents are
28 available, but most agencies in the UK use sodium chloride, which can be applied to
29 roads as a liquid or solid, depending upon the conditions (Blomqvist and Johansson,
30 1999). The salt may be relatively pure NaCl, or mixed with grits and sands, and
31 possibly an anti-caking agent such as sodium hexacyanoferrate (II) (Ohno, 1990).

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Email: sg507@york.ac.uk (S.M. Green)

About the Potomac Drinking Water Source Protection (DWSP) Partnership

- Voluntary regional organization
 - Water Utilities
 - State and Federal Agencies
- 20 Signatory Members
- Administrative support by ICPRB
- Focus on Drinking Water
- Impact of De-Icing is a Priority Issue



Fairfax Water

Questions?

